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ChemActivity 10 1. Na has 11 protons in its nucleus and Ne only has 10 protons. Therefore, the 1s electrons will be held more tightly in Na. 2. It would require more than 0.50 MJ/mole because Mg+1 and Na are isoelectronic and Mg has an additional electron in its nucleus. ChemActivity 11 1. Kr 2. C<Ne<Zn<Ba<Gd<Pt 4.

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ChemActivity 1 The Nuclear Atom 3 Critical Thinking Questions 11. "How many protons are found in  $^{22}\text{C}$ ?  $^{13}\text{C}$ ?  $^{12}\text{C}$ ?  $^{13}\text{C}$ ?  $^{12}\text{C}$ ?" 122. 13How many neutrons are found in  $^{12}\text{C}$ ?  $^{13}\text{C}$ ?  $^{12}\text{C}$ ?" 133. 13How many electrons are found in  $^{12}\text{C}$ ?  $^{13}\text{C}$ ?  $^{12}\text{C}$ ?" 4. Based on your answers to CTQs 1-3, what do all carbon atoms (and ions) have in common? 5.

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